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SYNTHESIS AND CHARACTERISATION OF CHALCONE BASED COPOLYESTERS AND THEIR ANTICANCER ACTIVITY

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ABSTRACT: One of the most prevalent causes of mortality among women is breast cancer. There are chemopreventive and anti-cancer properties shown by chalcone. Two copolyesters employing different chalconediols as a monomer are part of a proactive strategy in the battle against breast cancer. PG and the solution polycondensation technique were used to synthesize aliphatic and aromatic acid chlorides. Viscosity and solubility tests defined the two copolyesters. The copolyesters' structures were determined using FT-IR, ¹H NMR, ¹³C NMR, and MALDI. We used the MCF-7 Cell Line to study the monomer and copolyesters' anticancer activities, and the VERO Cell Line using the MTT test to study their cytotoxic effects. Graph Pad Prism 7 and ED50 Plus v.1.0 were used to compute the IC₅₀ values. Chalcone MHEP is cytotoxic, and the Selectivity index shows that chalcone HHEP and the two copolyesters are effective anticancer agents. Chemotherapy may target these copolyesters either alone or in conjunction with cancer drugs since their selectivity indices are greater than 2.

Keywords: Chalconediols, Copolyesters, MTT Assay, IC₅₀, Selectivity index

INTRODUCTION

Cancer refers to any malignant growth or tumor caused by abnormal and uncontrolled cell division. Breast cancer is one of the most common cancers in women. Death from cancer worldwide are projected to continue rising, with an estimated 131 million death in 2030. Chalcones are a class of natural compounds that widely exist in a variety of plant species. They are commonly known as yellow pigments in flowers and are also widely distributed in various parts (roots, rhizomes, heartwood, leaves, flowers and seeds)¹. Chalcone derivative have shown anti-cancer, antimalarial, anti-inflammatory², anti-bacterial, anti-filarial, antifungal and anti-oxidant activity³. Concerning the chemical structural level, these compounds are open-chained molecules, in which the two aromatic rings are joined by a three-carbon α , β -

unsaturated carbonyl system (1,3-diphenyl-2-prop-1-one). A large number of chemical structures have been studied, and it was reported that hydroxyl- derivatives of chalcones display marked anti-proliferative effects on cancer cells. These groups are likely necessary for the chemotherapy agents⁴. The principal aim of our work is the discovery of novel cytotoxic and anticancer agents. In the present study, we have synthesized two chalconediols by acid catalysed Claisen condensation method. Using those chalcone as monomers and varying acid chloride two copolyesters were synthesized to study their enhanced biological activity^{5,6,7}. The newly synthesized chalconediols and polyesters are further screened for their anticancer activity.

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MATERIALS AND METHODS:

Synthesis of Chalcone Diols: The two monomer chalcone diols namely (2E)-1-(4-hydroxy phenyl) 3-(4 hydroxy -3- ethoxy phenyl) prop-2- ene -1-one [HHEP] and (2E)- 1-(4 hydroxy-3- methoxy phenyl)- 3-(4 hydroxy -3- ethoxy phenyl) prop-2- ene -1 -one [MHEP] were synthesized and reported⁸.

Synthesis of HHEP: Dry HCl gas (generated as a result of reaction between concentrated sulphuric acid and dry crystals of NaCl) was passed through a well cooled solution of 3 g of 4-hydroxy-3-ethoxybenzaldehyde and 2.7 g of 4-hydroxyacetophenone in 100 ml of absolute ethanol

taken in a 250 ml RB flask under stirring conditions. The precipitate is poured into ice cold distilled water and allowed to settle, filtered, dried and recrystallised from methanol. Same method was adopted to synthesise MHEP and both were reported⁸.

Synthesis of Copolyesters:

Synthesis of PIOH: Taken 1g of HHEP in 100 ml RB flask. Added 10 ml of dry DMF stirred till it is dissolved, then added 0.3573 g of isophthaloyl chloride and 0.1586 g of oxalyl chloride, temperature was raised to 120 °C and maintained at 120 °C. Left the reaction to proceed for 12 h. After the reaction time, quenched in 50 ml of *n*-hexane, filtered, dried, powdered. It is then reprecipitated from methanol.

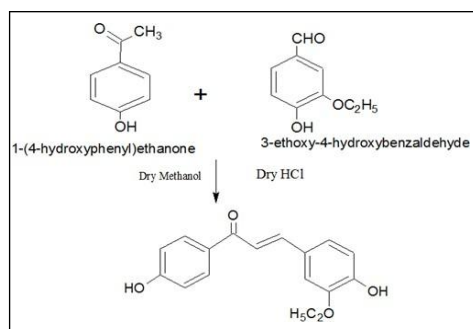


FIG.1: SYNTHESIS OF HHEP

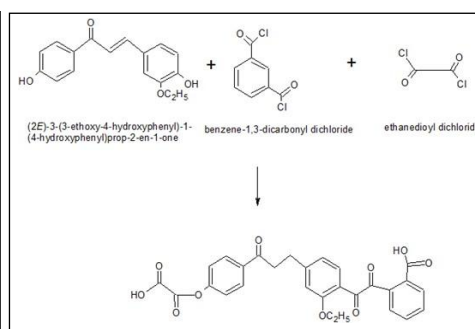


FIG.2: SYNTHESIS OF PIOH

Same method was adopted to synthesise PIOM and both were reported⁸. The structure of the copoly-esters was established by FT-IR, ¹H NMR, ¹³C

NMR and MALDI. Table 1 gives information regarding the monomers used in the synthesis of the two copolyesters, their respective copolyester code, percentage of yield and inherent viscosities.

TABLE 1: MONOMERS USED, COPOLYESTER CODE, PERCENTAGE OF YIELD AND INHERENT VISCOSITIES

Diol	Diacid chloride-I	Diacid chloride-II	Copolyester code	Percentage yield	Inherent Viscosity (cP)
HHEP	Isophthaloyl chloride	Oxalyl chloride	PIOH	78	1.2890
MHEP	Isophthaloyl chloride	Oxalyl chloride	PIOM	85	1.0578

Cytotoxic Studies: The anticancer activity of samples on MCF-7 cells were determined by the MTT assay cells (1x10⁵/well) were plated in 0.2 ml of medium/well in 96-well plates. Incubate at 5% CO₂ incubator for 72 h. Then, add concentrations of the samples in 0.1% DMSO for 48 h at 5% CO₂ incubator. After removal of the sample solution and washing with phosphate-buffered saline (pH 7.4), Viable cells were

determined by the absorbance at 540nm. Measurements were performed and the concentration required for a 50% inhibition of viability (IC₅₀) was determined graphically. The effect of the samples on the proliferation of MCF-7 cells was expressed as the % cell viability, using the following formula:

$$\text{Cell Viability (\%)} = \frac{\text{A}_{540} \text{ of treated cells}}{\text{A}_{540} \text{ of control cells}} \times 100$$

20 μl/well (5 mg/ml) of 3-(4,5-dimethyl-2-thiazolyl)-2,5-diphenyl-tetrazolium bromide (MTT)

viability = A_{540} of control cells $\times 100\%$
 phosphate-buffered saline solution was added. After 4 h incubation, 1 ml of DMSO was added. Using the GraphPad Prism, the IC_{50} values are calculated and compared with ED_{50} plus 1.0.

From this Selectivity index was found using the following formula: IC_{50} values of normal cell line

Spectral Studies: FTIR spectrum of the two copolyesters namely PIOH, PIOM were recorded by making use of Shimadzu FT-IR instrument. The

$$\text{Selectivity Index} = \frac{IC_{50} \text{ of normal cell line}}{IC_{50} \text{ of cancer cell line}}$$

FT-IR spectrum of the two copolyesters displayed distinctive absorption in the range of 1750-1768

RESULTS AND DISCUSSION: The two synthesized chalcone based copolyesters were characterized by viscosity measurement, solubility study and spectroscopic methods FT-IR, 1H and ^{13}C NMR and MALDI. Cytotoxic study was carried out using MTT assay. Solubility of the two copolyesters was determined in various solvents and reported ⁸. The outcome reveals that they are found to be highly polar

solvents, partially soluble in moderately polar solvents but absolutely insoluble in least polar solvents. The inherent viscosity value of the two copolyesters PIOH, PIOM was determined in DMAc solution at 30 °C using Ubbelohde Viscometer and were found to be 1.0189 and 1.0632 dL/g respectively as presented in Table 1 reveals that these copolyesters have high molecular weight cm^{-1} owing to ester C=O stretching frequency.

Nuclear Magnetic Resonance (NMR) spectroscopy is useful for characterizing the polymer samples. The 1H and ^{13}C NMR spectra of the two copolyesters were recorded with BRUKER AV III 500 MHz NMR instrument in DMSO-d6 solvent and TMS as internal standard. The aromatic protons were observed in the range 7.0 to 8.2 ppm. The vinylic protons attached to carbonyl carbon are observed in the ranges of 6.6 to 6.9 ppm. The vinylic protons attached to phenyl rings are observed in the ranges of 4.3 to 4.5 ppm. The methoxy protons in the chalcone moiety were observed in the range 3.1 to 3.4 ppm. The methylene protons observed in the range of 2.2 to

2.9 ppm and the methyl protons in 1.3 to 1.7 ppm.

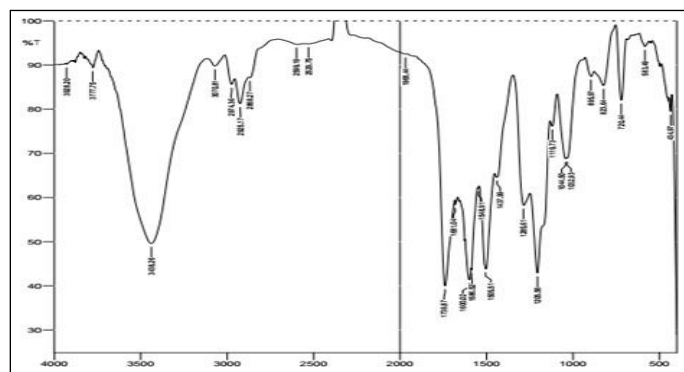


FIG.3: FT-IR SPECTRUM FOR PIOM

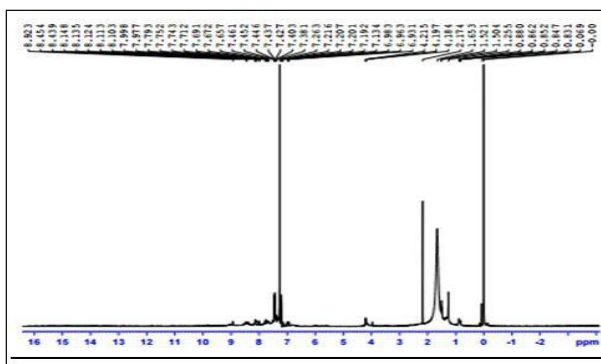


FIG.4: 1H NMR SPECTRUM FOR PIOM

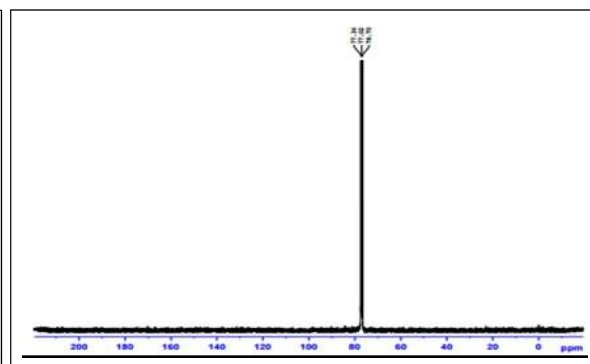


FIG.5: ^{13}C NMR SPECTRUM FOR PIOM

MALDI: MALDI-Mass spectrum was useful in finding the molecular weight of the polymeric samples. MALDI -Mass spectrum of PIOM is shown in the **Fig.6**. The Molecular weight of the

polymer was found to 877.242 experimentally, which is in nearly in agreement with the theoretical molecular weight.

***In-vitro* Cytotoxic Effect of Chalcones:**

Anticancer activity of chalcone might be due to molecular alteration such as induction of apoptosis, DNA and mitochondrial damage, inhibition of angiogenesis, tubulin inhibition,

kinases inhibition and also drug efflux protein activities⁹. Aromatic chalcones with hydroxyl group, methoxy group on A ring at position 2 or 3 are considered as lead compounds for generation of new potential anticancer drugs¹⁰. When the two monomers are compared for their anticancer activities, the chalcone diol HHEP was found to be good anticancer agent and MHEP as cytotoxic effect. When the two copolyesters are compared for their anticancer activities both are found to be good anticancer agents because the aromatic ring has substituent alkoxy groups at the chalcone backbone.

TABLE 2: IC₅₀ AND SELECTIVITY INDEX VALUES OF MONOMER ON VERO AND MCF-7 CELL LINE (GraphPad Prism Version 7)

Monomer Code	GraphPad Prism Version 7		
	IC ₅₀		Selectivity index
	VERO	MCF-7	
MHEP	2.499	2.349	1.0639
HHEP	12.8	2.754	4.6479

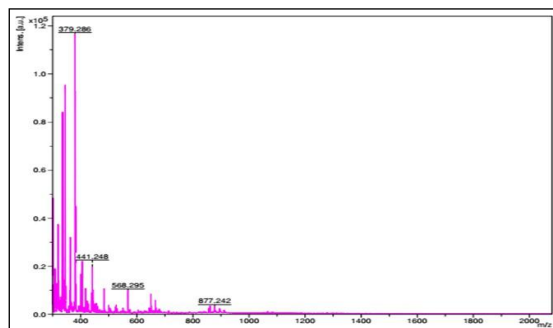


FIG.6: MALDI FOR PIOM

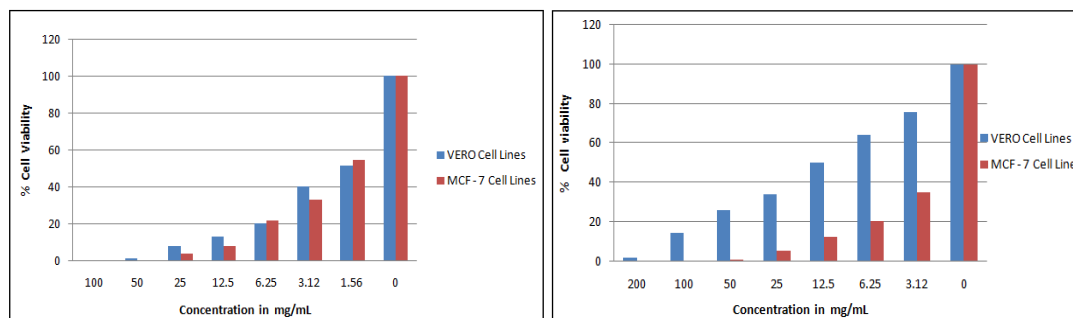
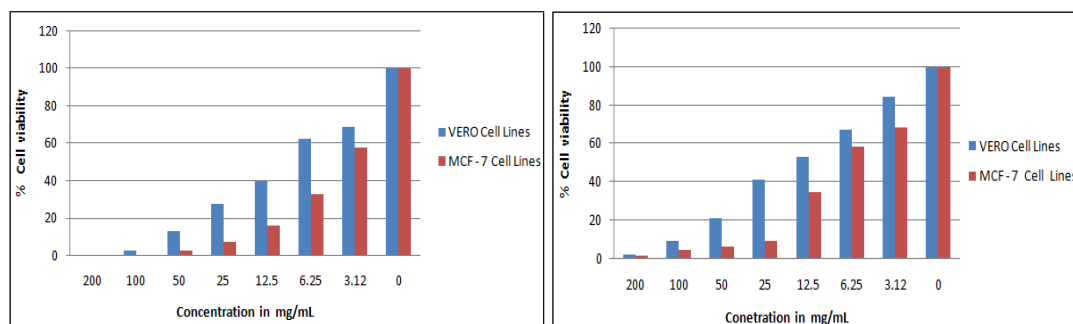


FIG.7 COMPARATIVE ACTIVITY OF MHEP ON VERO AND MCF-7 CELL LINES



**FIG.9:COMPARATIVEACTIVITYOFPIOH ON VERO AND MCF-7 CELL LINES
TABLE3:IC₅₀ANDSELECTIVITYINDEXVALUESOF COPOLYESTERSONVEROANDMCF-7CELLLINE(GRAPH
PAD PRISM VERSION 7 AND ED₅₀ PLUS V 1.0)**

Copolyesters Code	GraphPadPrismVersion7			ED ₅₀ PlusV1.0		
	IC ₅₀		Selectivityindex	IC ₅₀		Selectivityindex
	Vero	MCF-7		Vero	MCF-7	
PIOH	9.351	4.387	2.1315	8.73	3.24	2.69
PIOM	14.71	7.286	2.0189	14.99	6.274	2.38

Literature survey indicates that when the selectivity index is greater than two for a material under investigation it can be good anticancer agents¹¹. The Selectivity index of the monomer HHEP and the copolyesters PIOH and PIOM are found to be more than two. Hence they can be targeted at the Breast Cancer Cell Lines as good anticancer agents.

CONCLUSION: Two chalcone (HHEP and MHEP) were synthesized and characterized. These chalcone as monomers diols were used and two copolyesters were synthesized using different diacid chloride namely oxalyl chloride and isophthaloyl chloride by polycondensation method. These are characterized and evaluated for their anticancer activity. The selectivity index reveals that one of the monomer (HHEP) and two copolyesters are potent anticancer agents. Hence they can be used as anticancer agents. This study can be further extended on drug conjugates targeting MCF-7 Cell Line.

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